Pastry is like chemistry: just you can lick the spoon

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Context

• **Pastry = chemistry** (≠ you can eat the product)

Follow an experimental protocol, mix ingredients in a certain order in a certain manner to obtain different products

Egg + Sugar + Flour + Milk + Butter

same ingredients can give very different products depending on quantities, state and order of introduction



• Date : December 6th



St Nicholas' candies and chocolates







Role of Eggs



- 90% H2O
- 10% Proteins (mainly ovalbumin)
- Traces of minerals and vitamins



- 50% H2O
- 35% Lipids (lecithin)
- 15% Proteins

Beaten Egg Whites

Classic beaten egg whites:

Egg White X g at Room T

- Whip EW (increase speed progressively)
- Whip till fluffy foam

Modified beaten egg whites:

Egg White X g at Room T

- + sugar water (X g H2O + 0,15 X g sugar)
- Whip EW (increase speed progressively)
- Add progressively sugar water
- Whip till fluffy foam

65 g EW + 60 g H2O +10 g sugar \rightarrow ~ 1000 cm³ **Volume 15 fold increase**











65 g EW = ~ 2 EW

Beaten Egg Whites: What happens



Initial shape : wool ball

Ovalbumin = long chains containing hundreds of amino acids



Hydrophilic parts and hydrophobic parts in protein



Beaten Egg Whites: What happens



- Start with small whipping speed to introduce air bubbles and increase whipping speed to decrease air bubble size as smaller bubbles are easier to stabilize
- Bonds between denatured proteins increase foam stiffness
- Add of H₂O dilutes proteins and increases amount of bubbles that can be stabilized





- **Strictly avoid egg yolk** in EW and beat EW in clean container as lipids compete with proteins for surrounding air bubbles and lipids do not stabilize air bubbles
- Do not add salt but eventually a few drops of acid (vinegar, lemon juice)



AC2 ajout acide, concentration of H3O+ increases, decreasing the deprotonation of acid functions of proteins => no charging and less repulsion Alie Christelle; 03-12-22

Stability of Beaten Egg Whites

No endless stability because H₂O runoff along proteins and leakage



3 types of Meringues

French Meringue:

Egg White X g at Room T + 2 X g superfine sugar

- Whip EW (increase speed progressively) and add sugar progressively while whipping
- Whip till stiff glossy peaks

Swiss Meringue:

Egg White X g at Room T + 2 X g superfine sugar

- Whip EW+sugar over double boiler till 45-50°C
- Whip to cool down till stiff glossy peaks

Italian Meringue:

Egg White X g at Room T + 2 X g superfine sugar + 0,85 X g H2O

- Heat H2O+sugar till 118°C
- Whip EW when sugar syrup around 90°C
- Add hot syrup to EG while whipping
- Whip to cool down till stiff glossy peaks



Meringue : What happens



Sugar stabilizes EW in two ways :

- Sugar inserts between proteins to limit excessive bonding between proteins
- Sugar is hygroscopic and retains water inside meringue

Italian meringue : hot syrup induces coagulation of EW and part of H_2O in foam vaporizes but vapor is trapped in EW structure $\Rightarrow V \uparrow \uparrow \uparrow$

As mixture cools down, sugar crystallizes \Rightarrow glossy aspect of meringue

Stability of Meringues



Stability strongly influenced by heating during EW whipping

After 1h30 cooking at 100 °C







French mer.

Swiss mer.

Italian mer.

Applications:



Role of Egg Yolk

- Moistening and binding agent ۲
- **Traps air bubbles** ۲

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Egg Yolk + Sugar \Rightarrow white preparation (V \uparrow)
Egg Yolk + hot Sugar Syrup \Rightarrow white airy preparation (V \uparrow\uparrow\uparrow\uparrow)
    \Rightarrow very airy mousses
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Proteins of Egg Yolk are involved in **coagulation** of preparations **during heating** ۲



Role of Sugar

- Saccharose + $H_2O \rightarrow Glucose + Fructose$
- Flavor (sweet taste)
- Color



When heated, sugar **caramelizes** (set of simultaneous and successive reactions C₁₂H₂₀O₁₀ dianhydride de glucose : caramel smell) or undergoes a **Maillard reaction** (browning) if proteins are present (eg: characteristic brown color of cake crust)



- Partial dehydration of cake exterior (T >150°C):
 Maillard's reaction between sugar + protein (amino acids)
 ⇒ aromatic compounds (brown smelly crust)
- Cake interior <100 $^{\circ}C \Rightarrow$ no browning

Volume

Add volume to Egg white, Egg Yolk, Butter while whipping Add volume to baking by increasing T of protein coagulation

Softness

Increase softness by hindering gluten formation

Humidity

Sugar hygroscopic, retains H_2O in baking \Rightarrow softness

Crunchiness and Conservation

Role of Flour

• Proteins

Form gluten when in contact with liquid => elasticity and favors dough raise Gluten formation can be controlled by addition of Sugar and/or Fats Initially proteins form "wool balls" due to intramolecular hydrogen and disulfide bonding and after dough kneading : proteins unfold and align





Intermolecular disulfide Intramolecular disulfide

Non covalent bonding AC1

➢ Give taste by participating to Maillard's reaction with sugar Sugar + amino acid → glycosilamine + H₂O isomerization of glycosilamine → ketosamine numerous reactions of ketosamine → many products



FURANONES sweet caramel burnt

- **Starch** (= carbohydrate = 70 % of flour weight)
 - Responsible of H₂O absorption in dough during baking T > 50-70°C : Hydrogen bonding between starch molecules and H₂O, H₂O penetrates starch grains and make them swell (gelatinization)
 Helps to keep cake structure after baking

| AC1 | a) protéines associées par des liaisons disulfures intermoléculaires conférant une certaine élasticité à la pâte. |
|-----|--|
| | b) protéines associées par des liaisons non covalentes intermoléculaires conférant une certaine viscosité à la pâte. |
| | Alie Christelle; 03-12-22 |

Role of Raising agents during heating

- 1. Air slight effect due to low dilatation
- **2.** H_2O liquid \Rightarrow gas volume >>>
- 3. CO₂ : Sodium bicarbonate and/or Baking powder (tip: mix with flour before adding)
 - Sodium bicarbonate

 $2 \text{ NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$

(Problem Na₂CO₃ reacts with lipids by saponification, better use in combination with acid)

• Baking powder = sodium bicarbonate + acid + cornstarch

 $T > 50^{\circ}C$ NaHCO₃ + R-H \rightarrow CO₂ + R-Na + H₂O



Role of Fats

• Cake texture different depending on butter texture

hard \Rightarrow brittle dough (shortbread), soft \Rightarrow airy soft dough (sponge cake) or melted \Rightarrow dense soft dough (pancakes)

- Flavor vector
- Tenderize

Fats induce shortening of gluten network by protein coating

Aerate

Solid Fats (butter) incorporate air in preparation while whipped with sugar

Delay staling

Staling = crumb hardening due to structural change of starch

• Make puff pastry

Butter layers melt during heating, is incorporated in dough and leaves space for expanding

gas





Application : Cakes

- **Butter + sugar** incorporation of air through friction of sugar crystals with butter
- + egg at RT (as cold eggs \rightarrow less volume) |AC4|
- Flour sieved with baking powder (→ less clumps and better mixing)
- Flour chocolate chips, fruits, nuts otherwise they end up at the bottom of the mold.
- Flour and liquid added alternatively starting and finishing with flour

1st add of flour no gluten formation as proteins completed coated by lipids, once liquid az added, every non-coated protein will generate gluten

Alternate adding enables to generate enough gluten to give structure to cake without making as it compact and heavy. **Do not mix too long at that stage.**

- Color of mold has importance : dark mold absorbs more heat so cake bakes faster
- **Oven T** affects aspect and texture: the more eggs, milk and flour are long to coagulate, the more the bubbles have time to swell the cake.
- **Do not open oven** during first 15' as cake structure is not yet established

AC4 (increase whipping rate progressively to maximize volume) Alie Christelle; 05-12-22

Application : Pastry cream

Few Ingredients : Egg Yolk + sugar + milk + cornstarch but ... not so simple !

- Heat Milk (whole milk \rightarrow smoother and more flavored cream) + Flavor (eg vanilla)
- Egg Yolk + Sugar : whip immediately (if not, clumps), not necessary to whiten mixture
 + Cornstarch
- Add hot milk progressively to mixture (if not, egg will coagulate \rightarrow clumps)
- Cook on heat again carefully : mix continuously and do not overheat

why: - mixing induces good heat diffusion and good intercalation of starch that gelatinizes aaeerrrr and proteins that form network

- overheat induces coagulation of egg yolk
- Cook till bubbles appear and continue 1' while mixing

otherwise: - enzyme of starch not inactivated \rightarrow cream texture will not be maintained

- flour-taste due to insufficient cooking of starch



AC3 aussi custard cream Alie Christelle; 04-12-22

Alternatives

Flour :

- Replace 60% by almond powder (or hazelnut powder)
- Replace 50% by cornstarch
- Replace 25% by ground flaxseed

Butter :

- Replace by oil (80% oil 20% H2O)
- Replace by oleaginous puree or white cheese/mascarpone
- Remove up to 25% of fats
- Lighter : replace by apple compote or zucchini puree or avocado puree

Egg:

- replace by apple compote, yogurt or mix cornstarch+H2O
- Replace by ground flax/chia seed+water

Egg white :

• Replace by chickpeas water

Sugar :

- Replace by honey, agave syrup, maple syrup, stevia, date puree
- Replace part of sugar by equivalent amount of apple compote and remove ¼ part of liquid
- Decrease sugar by 20-25% possible

Milk :

• Replace by vegetal milk

Thank you for your attention

It's your turn to use your creativity



